Gold Prospecting by Pan



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COVER PHOTOGRAPH Gold nugget from the Tipuani River

(Bolivia), from the author's collection.

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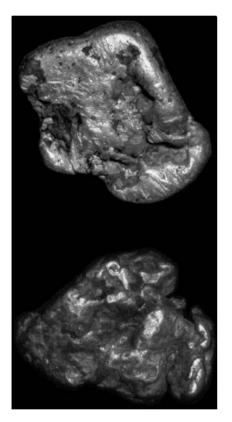
PREAMBLE

Since the 1990s, prospecting for gold and oil — or «mineral prospecting» — has become one of the most active economic areas. This frenetic activity is directly related to increasing demand from emerging market economies, like China, Brazil and, more traditionally, India, and that of Western economies, which consider gold a value-refuge. This has led, given its rarity of less than 4 milligrams per tonne of rock, to extreme pressure on the planet Earth, affecting its sustainability. In short, the richer its inhabitants the greater the demand.

The search process and subsequent extraction is known as «mineral exploration», defined as «the part of Geology which is interested in the search for minerals, rocks and energy resources with economic potential». Within mineral exploration, the most historical, traditional and cheap method would be alluvial prospecting or «prospecting by pan». This method and the associated technique — «panning» — is directly concerned with finding and assessing heavy metal deposits that indicate the existence of placer or residual deposits, and thus indirectly with determining the location of mineral and, therefore, geochemical anomalies. This requires the sampling of rivers and streams, which can lead to source or primary deposits of all types of minerals or also show up other companion minerals known as indicators or pathfinders.

This technique, logically used in the early stages of exploration, is not only employed for gold but also for substances known in their raw form as heavy minerals of commercial interest, such as platinoids and gems in the broad sense, including diamonds and technological and strategic minerals like titanium, zirconium, hafnium, rare soils, tin, tungsten and niobium-tantalum, the latter mined as «coltan».

A pan is a simple, cheap, easily obtained and easy-to-use recipient shaped like a dish, pan, bowl or basin, without sensors or electronics, which in the 21st century remains the most effective and efficient tool in mineral exploration. Some of its forms, as can be seen in this book and the accompanying video, are the result of empirical



knowledge, materials close to hand, and the intuitive application of fluid mechanics, which is to say fluvial dynamics. Gold pans are used by millions of people worldwide for subsistence artisanal or small-scale mining, and are typically used in the Western world in the pursuit of panning as a hobby or sport.

CHARACTERISTICS OF GOLD

Gold, chemical symbol Au, occupies position 79 on the periodic table, between platinum and mercury, and belongs to Group 11 together with copper and silver. It is a relatively soft metal, ductile and malleable, and a good conductor of heat and electricity. Both gold and the other two elements in Group 11 show little similarity to the alkaline metals of Group 1.

Atomic number: 79 Atomic weight: 196.97 Colour: metallic yellow Hardness: 2.5–3.0 Isotopes: 197 and 198

Electrical resistivity: 22.14 n Ω m Thermal conductivity: 318 W·m $^{-1}$ ·K $^{-1}$

Melting point: 1.063 °C Density: 19.3 g/cm³

Clarke value or average content

of the earth's crust: 2 mg/t or 2 µg/kg or 2 ppb

Gold is a fairly widespread chemical element in the earth's crust but at very low concentrations, of around 4 mg/t (0.004 grams per tonne of rock). Comparing this to another element such as copper, the average content of the latter is 23,000 times that of gold at around 47 grams in each tonne of rock. This proportion is not consistent with their monetary values, as gold is only worth 3,200 times more than copper.

Another characteristic of gold is its high density relative to water. A litre of water weighs one kilogram, while the same volume of gold would weigh 19 kilograms and 300 grams.

In the natural environment gold occurs as a native metal or alloyed with other metals, principally silver (Ag), copper (Cu), antimony (Sb), bismuth and metals of the platinum group. When the Ag content is higher than 20% it is called electrum. [Table 1]

One of the peculiarities of gold, due in part to its widespread diffusion in the earth's crust compared to other metals of economic interest, is the large number of types of mineral deposits it can form, as either an element or mineral base, even as a by-product of other mineral ores.

Table 1. Some gold minerals

Name	% content	Formula
Gold	100	Au
Electrum	55-80	Au-Ag
Maldonite	65	Au ₂ -Bi
Calaverite	39.5	(Au, Ag)Te ₂
Krennerite	39.5	
Sylvanite	24.5	(Au, Ag)Te ₄
Petzite	18-25	(Au, Ag) ₂ Te
Nagyágite	6-13	$Au_{2}Pb_{14}Sb_{3}Te_{7}S_{17}$
Fischerite	18-25	5Ag ₃ AuSe ₂

GOLD IN NATURE

Most authors agree that the abundance of gold in Nature is equivalent to an average of 4 milligrams per tonne of earth, with highest levels of content occurring in basic and ultrabasic rock. [Table 2]

Among sediments the most significant are the modern detrital formations, sometimes called «placers», as well as the ancient *rand* conglomerates in South Africa and black shale.

Table 2. Lithospheric gold content, modified after Boyle 1979, in ppb or mg/t

Source	Gold
Oceanic crust	3.5
Continental crust	3.5
Terrestrial crust	3.5
Crystalline rock	3.6
Sedimentary	5.1
Ultrabasic	11.4
Basic intrusive	23.0
Basic extrusive	17.0
Intermediate intrusive	7.5
Intermediate extrusive	12.9
Acidic intrusive	11.4
Acidic extrusive	3.7
Alkaline	3.4

Source	Gold
Quartzite	31.7
Slate and phyllite	2.2
Gneisses and granulite	3.1
Amphibolites	7.1
Schist	18.6
Marble	13.2
Corneal	6.4
Eclogite	3.6
Skarns (Tactites)	8.6
Detrital	57.0
Greywacke	13.2
Claystone	8.0
Black shale	132.0
Tuff	6.9
Limestone	7.0
Gypsum evaporite	20.8
Salts	23.8
Chert	16.7
Phosphorite	7.2

Gold in water

Au is widely spread in groundwater and surface water, natural springs, sea water and so on, with an average fresh water content of 0.03 μ g/l (although many authors believe it to be significantly lower) and 0.012 μ g/l in marine waters; around 0.53 μ g/l Au in geothermal systems.

There are marked variations above and below this average, if we look at contents in specific types of water body, associated with different geological terrain. Camacho (1988) observes contents ranging from 0.02-0.06 ppb in unfiltered water from the gold district of Penedono (Portugal) and 0.002-0.01 ppb in filtered water from the same district. [Table 3]

Table 3. Gold content in different water environments (µg/l)

Sample type	No.	Range of Au	Observations
River-creek	36	<0.002-0.010	11 % between 0.008-0.010
Spring	6	<0.002-0.013	33 % between 0.009-0.013
Mine gallery water	12	<0.002-0.014	41 % between 0.007-0.01

Another example performed by the same author in a creek near the gold and antimony mine at Los Machos (Bolivia) showed that at a single point, where ten samples (LM1.1-LM1.10) were taken at a sampling rate of 1 litre of water every 20 seconds (stream flow 3 m³/s), the gold values of different unfiltered samples are completely erratic, ranging from 0.05-2.4 mg/l. The high values of Au, arsenic (As), zinc (Zn) and antimony (Sb) observed in the first of the ten samples suggest the presence of gold particles. [Table 4]

Table 4. Content of gold, As, Ba, Sb, Br and Zn in creek water from a sub-tropical zone in Bolivia (µg/l)

Sample	Au	As	Sb	Zn
LM1.1	2.40	4	9	60
LM1.2	0.45	2	3	30
LM1.3	0.24	2	2	40
LM1.4	0.48	2	2	20
LM1.5	0.12	2	2	40
LM1.7	0.09	2	3	30
LM1.8	0.06	2	2	30
LM1.9	0.05	2	2	50
LM1.10	0.08	2	2	40
«LM2»	0.42	2	4	50
«LM3»	0.18	1	2	40
«LM4»	0.16	1	2	60
«LM5»	0.05	1	2	70

Gold in vegetation

Numerous works from the mid-20th century tell us of the gold content in plants and animals.

Gold concentrations in humus, matter that is halfway between soil and vegetation, are more evenly spread than in alluvial, colluvial and glacial deposits. The causes of gold presence in humus can be related to the gold retention capacity of organic matter, as well as a reduction effect or gold absorption by plants, mainly cyanogenic plants that, in order to dissolve vital elements such as Zn, produce cyanides that also dissolve gold (Shacklette 1970).

Table 5. Gold concentration in some plant species (mg/t) of the Iberian Peninsula (Viladevall 2008).

Plant species	Location	No.	Gold ppb
Alnus glutinosa	Planoles gold mine, Central Pyranees	6	12.5
Arbutus unedo	Montseny	1	1.2
Buxus sempervirens	Martorell - Vallirana	20	2.8
Capparis spinosa L	Almeria	1	3.0
Cistus monpeliensis	Alluvial river Llobregat, Sant Feliu	1	5.5
Cistus salvifolius	Breda zinc mine, Montseny	1	2.5
Corilus avellana	Auriferous alluvial river Segre, Balaguer	1	6.7
Erica multiflora	Martorell - Vallirana	20	1.9
Genista anglica	Planoles gold mine, Central Pyranees	1	13.5
Juglans regia	Auriferous alluvial river Segre, Balaguer	1	4.4
Juniperus comunis	Planoles gold mine, Central Pyranees	1	8.9
Medicago sativa	Basaltic soil, Olot	1	11.4
Pinus pinea	Planoles gold mine, Central Pyranees	1	22.0
Pinus nigra	Victoria Mine, Central Pyranees	45	4.2
Pinus pinea	Maresme	35	3.2
Pinus holepensis	Martorell - Baix Llobregat	1	2.0
Platanus acerifolia	Average content in Barcelona	54	47.4
Platanus acerifolia	Caldetes	1	10.0

Plant species	Location	No.	Gold ppb
Platanus acerifolia	Auriferous alluvial river Segre, Balaguer	1	2.2
Populus nigra	Planoles gold mine, Central Pyranees	10	19.3
Populus nigra	Alluvial river Ebro, Flix - Ascó	12	13.4
Quercus faginea	Planoles gold mine, Central Pyranees	19	5.9
Querqus suber	Breda zinc mine, Montseny	1	3.2
Quercus ilex	Maresme	18	2.4
Rubus ulmofolius	Planoles gold mine, Central Pyranees	1	6.5
Salix fragilis	Auriferous alluvial river Segre, Balaguer	1	1.8
Thymus vulgaris	Almeria	1	2.4
Cerezo	Planoles gold mine, Central Pyranees	1	4.6
Ciruelo	Alluvial river Llobregat, Sant Feliu	3	1.3
Manzano	Alluvial river Llobregat, Sant Feliu	4	2.6
Melocotonero	Auriferous alluvial river Segre, Balaguer	3	1.9
Nogal	Auriferous alluvial river Segre, Balaguer	3	4.4
Peral	Alluvial river Llobregat, Sant Feliu	3	1.6
Pimiento rojo	Azogue Valley, Murcia - Almeria	1	5.4